



# Electronics gases Going green

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The electronics industry has, since the 1990s, depended greatly on the use of nitrogen trifluoride (NF<sub>3</sub>) for its process chamber cleaning needs. When initially introduced it provided a faster way of cleaning a chamber than previous methods using sulfur hexafluoride (SF<sub>6</sub>), tetrafluoromethane (CF<sub>4</sub>) and hexafluoroethane (C<sub>2</sub>F<sub>6</sub>). In the past six years, however, we've seen a change in attitudes towards NF<sub>3</sub> and its predecessors thanks to the growing understanding of NF<sub>3</sub>'s environmental impact.

From the declaration by Michael Prather in 2008 that NF<sub>3</sub> is the Greenhouse Gas missing from Kyoto, and the report from the Scripps Institute showing the increase in measured NF<sub>3</sub> concentrations in the upper atmosphere, followed by its addition to the UNFCCC agreement in 2011, the world has seen growing momentum around the movement to significantly reduce NF<sub>3</sub> emissions. With the commercial availability of fluorine (F<sub>2</sub>), a faster, lower cost and simpler method of cleaning manufacturing chambers that has zero global warming potential (GWP) compared to NF<sub>3</sub>'s 17,200 GWP100 rating is now available.

Linde has been the industry leader driving the move from NF<sub>3</sub> to F<sub>2</sub> in the electronics industry and has pioneered safe and reliable fluorine manufacturing systems and delivery techniques. Due to its advantageous chemical properties, F<sub>2</sub> reduces cost for electronics manufacturers, lessens their environmental impact, and creates major process efficiencies not available with NF<sub>3</sub>.

## Environmental impact

To minimise NF<sub>3</sub> process emissions, most manufacturers have installed high performance scrubbing systems. However, there still exists the risk of emissions during the whole production, transportation and use lifecycle. In 2008, Linde Electronics was quoted within the pages of *gasworld* saying, "You can mitigate something with a high global warming potential but if your alternative has zero global warming potential, it is fundamentally better." Supporting F<sub>2</sub> in the semiconductor and other electronics manufacturing industries is Linde Electronics delivering on that message.

NF<sub>3</sub> has a GWP 17,200 times greater than carbon dioxide (CO<sub>2</sub>). Molecular F<sub>2</sub> has no GWP value and with it being a very reactive molecule, there is very low concentration of the gas in exhaust streams, making it much more straightforward to abate. F<sub>2</sub> can also be generated on-demand and onsite from anhydrous HF and then consumed and abated at the same site, thus eliminating much of the risk of undesired emissions and lowering the total carbon footprint associated with the manufacturing, transport and disposal of NF<sub>3</sub> cylinders.

Across the entire value chain, in moving our customers to F<sub>2</sub> processes, Linde has eliminated 35,000 tonnes of CO<sub>2</sub> from the cleaning gas supply chain per year. However, the benefits of moving to F<sub>2</sub> go beyond its reduced environmental impact to

## Global Warming Potential of Greenhouse Gases (100 year GWP kg CO<sub>2</sub> eq)

F <sub>2</sub>	NF <sub>3</sub>	SF <sub>6</sub>
0	17,200	22,800

business process and operational cost, two areas that must remain core concerns in the competitive electronics industry.

## Cost

On a molecule basis, F<sub>2</sub> price can be very competitive with NF<sub>3</sub>, particularly at the large scales used for chamber cleaning in a large 300mm wafer fab.

There is also a reduction in the mass of gas required – 80kg of F<sub>2</sub> delivers the same amount of F-atoms as 100kg of NF<sub>3</sub>. You are not buying and transporting N atoms, which play no role in the cleaning process and in fact hinder the efficiency of the cleaning process.

Fluorine can also help improve factory productivity (thus lowering overall manufacturing costs) due to its up to 5x faster cleaning rate.

## Process benefits

Fluorine allows for significant throughput improvements over NF<sub>3</sub> in many cleaning processes. The NF<sub>3</sub>-based cleaning process typically uses a remote plasma source (RPS) to activate NF<sub>3</sub> but the F-N bond requires more energy to break than the simpler F-F bond and thus for a given RPS unit, higher flows of F<sub>2</sub> can be used compared to NF<sub>3</sub> which allows for shorter cleaning times.

Alternatively, for the same flow of F molecules, significantly lower power can be used with F<sub>2</sub> thus helping reduce power costs and also improving the reliability of the RPS units.

It has also been observed that the temperature of the reaction chamber typically increases during cleaning with NF<sub>3</sub>. This is not unexpected – the recombination mechanism for NF<sub>3</sub> can release significant amounts of heat energy but also more energy is required by the RPS unit to activate the same number of F atoms, and some of this power is dissipated as heat in the plasma. For a high temperature deposition process this effect may not be particularly significant, but for the newer low temperature process, accurate control of the process temperature is critical. Consequently, with NF<sub>3</sub>, additional recovery time may be required before subsequent wafers can be processed.

It is also possible in many cases to carry out the chamber cleaning with F<sub>2</sub> without the need for an RPS unit. The RPS unit is typically only present on the process tool because it is required for the cleaning step using NF<sub>3</sub>. Most plasma deposition processes use an in-situ method to generate the plasma and this same system can be used with F<sub>2</sub>, with no obvious adverse effects on the chamber hardware, thus simplifying the process tool.

In some cases the cleaning can also be carried out thermally without the use of plasma. All these benefits help reduce the amount of time the process tool is unavailable for production, thus improving overall productivity and helping to lower costs.

## A call to action


Linde Electronics has pioneered the use of molecular fluorine as a replacement for high GWP fluorinated cleaning or etching gases, such as NF<sub>3</sub> and SF<sub>6</sub>, which are routinely used in the manufacture of semiconductors and flat panel displays. It can also be used to replace chlorine trifluoride (ClF<sub>3</sub>) for similar cleaning applications,

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which, whilst not a greenhouse gas, can be hazardous to handle.

In order to fully realise all the benefits that F<sub>2</sub> can provide, some optimisation of the process may be required – and it is likely that for each OEM, unique process optimisation will be required to deliver the best results. This can be due to a number of factors – gas flow uniformity, plasma density, chamber pressure and temperature, for example.

However, whilst it may not be as simple as replacing one gas with another, the benefits that can be gained are significant. By implementing F<sub>2</sub> on the tools studied at a typical 300mm memory customer, tool availability improved by three weeks per year and there was a reduction in the mass of gas consumed by more than 2.5 tonnes, along with significant reductions in both power consumption and CO<sub>2</sub> equivalent emissions when the whole lifecycle of NF<sub>3</sub> is considered.

Due to the multiple significant benefits associated with the industry transition from NF<sub>3</sub> to F<sub>2</sub>, Linde Electronics remains committed to continue innovating in this area. There are now more than 30 Linde installations of fluorine cleaning processes globally across the semiconductor, display and photovoltaics (PV) industries, covering a wide range of OEM platforms and process types. We'll continue to expand that number and look for new ways of creating cost, environmental and process improvements within the electronics industry, while offering them a greener alternative. 

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